

## Chemical Waves and Chemical Oscillations

This interactive tool is helpful to observe the phenomena of Chemical Waves and Chemical Oscillations by using the Belousov-Zhabotinsky reaction (for more information about this reaction, please read the book “Untangling Complex Systems: A Grand Challenge for Science” by P.L. Gentili). Go to the wet laboratory, wear a white coat, gloves and safety glasses, and prepare the following solutions using deionized water as the solvent (see Figure 1):

- Solution A:  $\text{KBrO}_3$  0.6 M in  $\text{H}_2\text{SO}_4$  0.6 M.
- Solution B:  $\text{CH}_2(\text{COOH})_2$  0.48 M
- Solution C:  $\text{KBr}$  0.97 M
- Solution D: Ferroin (tris(1,10-phenanthroline) iron(III) sulphate) 0.025 M.



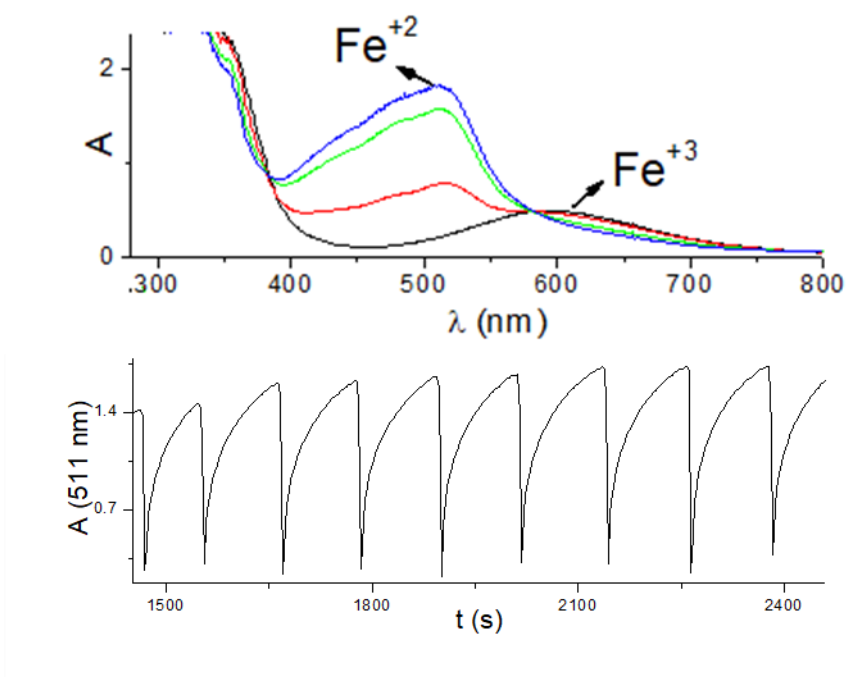
**Figure 1.** The glasses and the other tools required to perform this experiment.

Into a small Erlenmeyer flask, introduce 7 mL of solution A, 3.5 mL of solution B, and 1 mL of solution C. Close the container with a stopper and stir the solution with a magnetic stirrer. Bromate oxidizes bromide to bromine, and the solution looks brown (see Figure 2). The brown color slowly disappears because bromine reacts with malonic acid to form bromomalonic acid.



**Figure 2.** The preliminary steps for the BZ reaction, when bromine is formed and then consumed.

When the solution becomes transparent, add 0.5 mL of D and stir. Use a pipet to transfer 2.5 mL of the mixture into a cuvette having a tiny stir bar and pour the rest into a Petri dish (about 10 cm in diameter) to cover its surface uniformly. Place the cuvette in a UV-Visible spectrophotometer and maintain its solution under stirring. It is possible to record the spectrophotometric oscillations and determine the period of the chemical oscillations (see Figure 3).



**Figure 3.** UV-visible absorption spectra for the BZ reaction (on top) and time evolution of the absorbance at 511 nm (at the bottom).

Place the Petri dish on a sheet of millimeter graph paper and leave it quiescent for a while. Then, chemical waves will appear, as shown in Figure 4.



**Figure 4.** Chemical waves generated by the BZ reaction.